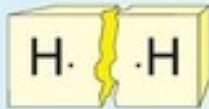


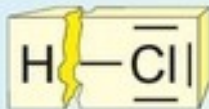
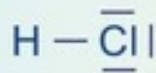
Oxidationszahlen

Wasserstoff



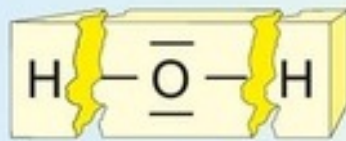
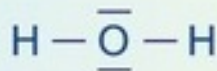
Wasserstoff: ± 0 $\overset{\pm 0}{\text{H}_2}$

Salzsäure



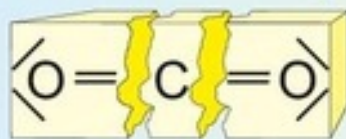
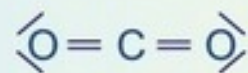
Wasserstoff: +I
Chlor: -II $\overset{+I}{\text{H}} \overset{-I}{\text{Cl}}$

Wasser



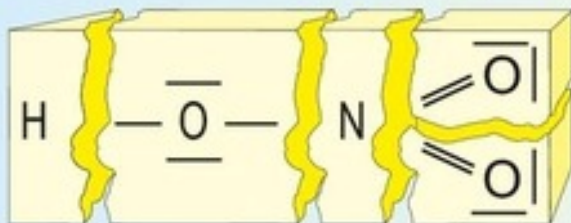
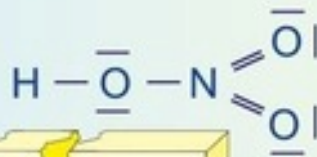
Wasserstoff: +I
Sauerstoff: -II $\overset{+I}{\text{H}} \overset{-II}{\text{O}} \overset{+I}{\text{H}}$

Kohlenstoffdioxid



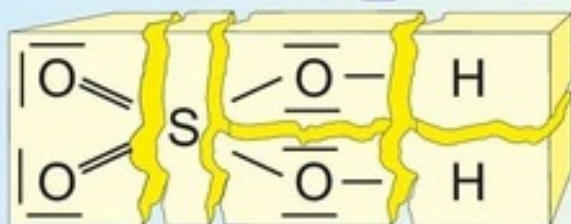
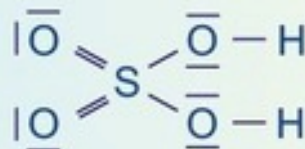
Kohlenstoff: +IV
Sauerstoff: -II $\overset{+IV}{\text{C}} \overset{-II}{\text{O}} \overset{-II}{\text{O}}$

Salpetersäure



Wasserstoff: +I
Sauerstoff: -II $\overset{+I}{\text{H}} \overset{-II}{\text{N}} \overset{-II}{\text{O}} \overset{-II}{\text{O}}$
Stickstoff: +V

Schwefelsäure



Wasserstoff: +I
Sauerstoff: -II $\overset{+I}{\text{H}} \overset{-II}{\text{S}} \overset{-II}{\text{O}} \overset{-II}{\text{O}} \overset{-II}{\text{O}} \overset{-II}{\text{O}}$
Schwefel: +VI